

⑫

EUROPEAN PATENT SPECIFICATION

⑬ Date of publication of patent specification: 18.10.89

⑭ Application number: 85109710.5

⑮ Date of filing: 02.08.85

⑯ Int. Cl.⁴: **C 08 F 220/04, C 08 F 246/00,
C 08 F 212/14, C 08 F 220/36,
C 07 C 125/065, C 09 D 5/02**

⑰ **Alkali soluble latex thickeners.**

⑱ Priority: 23.08.84 US 643370

⑲ Date of publication of application:
05.03.86 Bulletin 86/10

⑳ Publication of the grant of the patent:
18.10.89 Bulletin 89/42

㉑ Designated Contracting States:
BE DE FR GB IT NL SE

㉒ References cited:
US-A-4 384 096

㉓ Proprietor: DeSOTO, INC.
1700 South Mt. Prospect Road
Des Plaines Illinois 60018 (US)

㉔ Inventor: Shay, Gregory D.
15518 Rob Roy Drive
Oak Forest, IL 60452 (US)
Inventor: Eldridge, Emmejean
5410 West Thomas Street
Chicago, IL 60651 (US)
Inventor: Kail, James E.
1708 North Wilshire Avenue
Arlington Heights, IL 60004 (US)

㉕ Representative: Reitzner, Bruno, Dr.
Patentanwälte Dipl.-Ing. R. Splanemann Dr. B.
Reitzner Tal 13
D-8000 München 2 (DE)

EP 0 173 109 B1

Note: Within nine months from the publication of the mention of the grant of the European patent, any person may give notice to the European Patent Office of opposition to the European patent granted. Notice of opposition shall be filed in a written reasoned statement. It shall not be deemed to have been filed until the opposition fee has been paid. (Art. 99(1) European patent convention).

Description

Technical Field

This invention relates to latex polymers which are soluble in aqueous alkaline medium to provide thickeners for use in aqueous coating compositions, especially latex paints.

Background Art

Thickeners for aqueous systems are needed for various purposes, but they are particularly needed in aqueous latex paints to improve the rheology of the paint. Hydroxyethyl cellulose is a well known thickener for aqueous systems, but it has various deficiencies in that excessive amounts must be used and the rheology of the thickened system is inadequate. Various ethoxylated carboxyl-functional polymers which form alkali soluble thickeners are also known, but these have various deficiencies, including inadequate hydrolytic stability.

From the standpoint of alkali soluble thickeners which are carboxyl-functional emulsion copolymers, reference is made to U.S. Pat. No. 4,384,096 issued May 17, 1983. In that patent, 15—60 weight percent of a monoethylenic carboxylic acid is copolymerized in aqueous emulsion with 15—80 weight percent of a monoethylenic monomer, typified by a monovinyl ester, and 1—30 weight percent of a nonionic vinyl surfactant ester which is an ester of a nonionic ethoxylate surfactant with an acrylic or methacrylic acid. These copolymers are soluble in water with the aid of an alkali to form a thickener in the water solution.

Another prior patent of interest to this invention is U.S. Pat. No. 4,079,028 issued June 27, 1978 in which the thickener is a nonionic polyurethane of various types, but these do not provide their thickening characteristics as a result of alkali solubilization.

It has long been desired to provide superior thickeners for aqueous systems which are highly efficient, which better resist hydrolysis, and which provide better rheology. This is achieved herein by providing new alkali soluble anionic latex polymers which possess these desired characteristics.

Disclosure of Invention

In accordance with this invention, the alkali soluble thickener is an aqueous emulsion copolymer of:

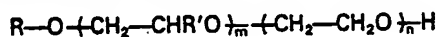
(A) about 20—70, preferably 25—55, weight percent of an alpha, beta-monoethylenically unsaturated carboxylic acid, typically methacrylic acid;

(B) about 20—80, preferably 30—65, weight percent of a monoethylenically unsaturated monomer lacking surfactant capacity, typically ethyl acrylate; and

(C) about 0.5—60, preferably 10—50, weight percent of a nonionic urethane monomer which is the urethane reaction product of a monohydric nonionic surfactant with a monoethylenically unsaturated monoisocyanate, preferably one lacking ester groups like alpha, alpha-dimethyl-m-isopropenyl benzyl isocyanate. The monohydric nonionic surfactants are themselves well known and are usually ethoxylated hydrophobes containing adducted ethylene oxide to provide the hydrophilic portion of the molecule. The hydrophobes are usually an aliphatic alcohol or alkyl phenol in which a carbon chain containing at least 6 carbon atoms provides the hydrophobic portion of the surfactant. These surfactants are illustrated by ethylene oxide adducts of dodecyl alcohol or octyl or nonyl phenol which are available in commerce and which contain about 5 to about 150, preferably 10 to 60 moles of ethylene oxide per mole of hydrophobe; and

(D) from 0 up to about 2 weight percent of a polyethylenically unsaturated monomer may be copolymerized into the copolymer, as is common in alkali soluble emulsion copolymers. These are illustrated by ethylene glycol diacrylate or dimethacrylate 1,6-hexanediol diacrylate or dimethylacrylate and diallyl benzene.

The preferred surfactants have the formula:



in which R is an alkyl group containing 6—22 carbon atoms (typically dodecyl) or an alkaryl group containing 8—22 carbon atoms (typically octyl phenol), R' is C₁—C₄ alkyl (typically methyl), n is an average number from about 6—150, and m is an average number of from 0—50 provided n is at least as great as m and n + m = 6—150.

It will be understood that urethanes can be made by various procedures, so the urethane reaction product used herein can be made in any desired fashion so long as the resulting product is essentially the same as that made by the reaction of the components named herein.

The thickeners of this invention possess structural attributes of two entirely different types of thickeners (those which thicken by alkali solubilization of a high molecular weight entity, and those which thicken due to association), and this may account for the superior thickener properties which are obtained herein. On the other hand, the ethoxylated monomer thickeners of this invention are not esters with unsaturated acids, as required in patent 4,384,096, and they do not have any of the diverse structures described by patent 4,079,028.

Considering the components which are copolymerized in aqueous emulsion in this invention, a large proportion of carboxylic acid monomer must be present, as previously defined. Various carboxylic acid

monomers can be used, such as acrylic acid, methacrylic acid, itaconic acid and maleic acid. Methacrylic acid is presently preferred. This large proportion of acid is essential to provide a polymeric structure which will solubilize and provide a thickener when reacted with an alkali, like sodium hydroxide.

The polymer must also contain a significant proportion, as previously defined, of a monoethylenic monomer which has no surfactant characteristic. The preferred monomers provide water insoluble polymers when homopolymerized and are illustrated by acrylate and methacrylate esters, such as ethyl acrylate, butyl acrylate or the corresponding methacrylate. Other monomers which can be used are styrene, vinyl toluene, vinyl acetate, acrylonitrile and vinylidene chloride. Nonreactive monomers are preferred, these being monomers in which the single ethylenic group is the only group reactive under the conditions of polymerization. However, monomers which include groups reactive under baking conditions may be used in some situations, like hydroxyethyl acrylate. The selection of these monomers is no different in this invention than it was in the prior art noted previously.

The monohydric nonionic surfactant component is subject to considerable variation within the formula presented previously. The essence of the surfactant is a hydrophobe carrying a polyethoxylate chain (which may include some polypropoxylate groups) and which is terminated with a single hydroxy group. When the hydroxy-terminated polyethoxylate used herein is reacted with a monoethylenically unsaturated monoisocyanate, as has been illustrated, the result is a monoethylenically unsaturated urethane in which a polyethoxylate structure is associated with a copolymerizable monoethylenic group via a urethane linkage. In this invention it has been found that alkali soluble polycarboxylic acid thickeners provide superior thickening action when urethane linkages are present even though the primary thickening mechanism is alkali solubilization of a polycarboxylic acid emulsion copolymer. At the same time, the hydrolytic stability of the urethane group is superior to the hydrolytic stability of the prior art ester group, and when preferred urethane monomers are used, this provides a further advantage in aqueous alkaline mediums as found in aqueous coating compositions, which are normally stored for long periods of time prior to use.

The monoethylenically unsaturated monoisocyanate used to provide the nonionic urethane monomer is subject to wide variation. Any copolymerizable unsaturation may be employed, such as acrylate and methacrylate unsaturation. One may also use allylic unsaturation, as provided by allyl alcohol. These, preferably in the form of an hydroxy-functional derivative, as is obtained by reacting a C_2-C_4 monoepoxide, like ethylene oxide, propylene oxide or butylene oxide, with acrylic or methacrylic acid to form an hydroxy ester, are reacted in equimolar proportions with an organic diisocyanate, such as toluene diisocyanate or isophorone diisocyanate. The presently preferred monoethylenic monoisocyanate is styryl, as in alpha, alpha-dimethyl-m-isopropenyl benzyl isocyanate, and this unsaturated monoisocyanate lacks the ester group so it forms urethanes which lack this group.

The aqueous emulsion copolymerization is entirely conventional and will be illustrated in the examples. To obtain an estimate of thickening efficiency, the latex product can be diluted with water to about 1% solids content and then neutralized with alkali. The usual alkali is ammonium hydroxide, but sodium and potassium hydroxide, and even amines, like triethyl amine, may be used for neutralization. The neutralized product dissolves in the water to provide an increase in the viscosity. In the normal mode of addition, the unneutralized latex thickener is added to a paint and then neutralized. This facilitates handling the thickener because it has a lower viscosity before neutralization. This procedure also makes more water available for the paint formulation.

The invention is illustrated by the following examples, it being understood that throughout this document, all proportions are by weight unless otherwise stated.

The following Example is illustrative of Examples 1-A through 1-L in Table I for the preparation of the nonionic urethane monomers of this invention.

Example 1

(Preparation of a Urethane Monomer)

To a one-liter glass reactor fitted with a thermometer, heating mantle, thermoregulator, stirrer, nitrogen sparge, and condenser including a Dean-Stark trap is charged 800.0 g of a 50 mole ethoxylate of nonyl phenol as a hot melt (Igepal® CO-970, a product of GAF, may be used). The reactor contents are heated, with nitrogen sparging, to 150°C and held for two hours while trace moisture is removed and collected in the Dean-Stark Trap (typically less than 1 g).

The reactor contents are then cooled to 80°C, the Dean Stark trap is replaced with a condenser, and the nitrogen sparge is switched to an air sparge for 15 minutes. With continued air sparging, 0.02 g methoxyhydroquinone inhibitor, 0.50 g dibutyl tin dilaurate catalyst, and 99.7 g of alpha, alpha-dimethyl-m-isopropenyl benzyl isocyanate (m-TMI, a product of American Cyanamide, may be used) are charged in order to the reactor. After a rapid initial exotherm which increases the reaction temperature about 8°C, the contents are heated to maintain 80°C for an additional two hours. The product is then cooled to room temperature.

The final product is a white wax in appearance with residual isocyanate content of 0.5% and with 98% of the original ethylenic unsaturation retained. This product is designated Example 1-D in Table I.

For purposes of comparison, the following Example is illustrative of Examples 2-A through 2-D in Table II for the preparation of an alkali-soluble thickener containing no urethane monomer. These latex thickeners

EP 0 173 109 B1

display poor performance because they possess low solution viscosity, have poor thickening efficiency, and provide poor flow leveling properties in latex paints.

Example 2

5 (Control)

To a three-liter flask equipped as in Example 3 hereinafter is charged 525.5 g deionized water. The water is heated to 80°C and purged with nitrogen for 30 minutes.

A pre-emulsion of monomers is prepared in a separate stirred container by charging, in order, 756.7 g deionized water, 81.6 g sulfonated octyl phenol ethoxylate containing about 10 moles of adducted ethylene oxide per mol of the phenol (Alipal EP-110 surfactant, a product of GAF, may be used) 367.0 g ethyl acrylate, 10 and 244.7 g methacrylic acid.

Under a nitrogen blanket, 145.0 g (10%) of the monomer pre-emulsion is charged to the reactor followed by 10.0 g of 5% sodium persulfate solution. The contents exotherm to about 85°C, and after cooling back to 80°C, the addition of the remaining pre-emulsion is started and continued over 2.5 hours until complete. The reactor contents are heated for an additional 30 minutes at 80°C to complete conversion 15 of monomer to copolymer and then cooled.

The product is low viscosity latex of solids content 28.9%, RVT Brookfield viscosity 28.5 mPa-s (No. 1 spindle at 100 rpm), pH of 2.8, and average particle size of 129 nm. Upon ammonium hydroxide neutralization to pH 9 of a 1% active concentration of this latex in water, a clear solution is obtained with 20 RVT Brookfield viscosity of only 60 mPa-s (No. 1 spindle at 10 rpm). This latex thickener is designated 2-C in Table II.

The following Example is illustrative of Examples 3-A through 3-U and 4—19 in Tables II and III for alkali-soluble thickeners prepared from the urethane monomers of this invention in Table I.

Example 3

25 (Preparation of an Alkali-Soluble Thickener with Urethane Monomer)

To a three-liter flask equipped with thermometer, stirrer, condenser, nitrogen inlet, thermoregulated water bath and monomer addition pump is charged 525.5 g deionized water. The water is heated to 80°C and purged with nitrogen for 30 minutes.

A pre-emulsion of monomers is prepared in a separate stirred container by charging, in order, 756.7 g deionized water, 81.6 g sulfonated octyl phenol ethoxylate containing about 10 moles of adducted ethylene oxide per mol of the phenol (Alipal® EP-110 surfactant, a product of GAF, may be used) and a monomer blend consisting of 305.8 g ethyl acrylate, 244.7 g methacrylic acid, and 61.2 g of the ethoxylated urethane monomer prepared in Example 1. 30

Under a nitrogen blanket, 145.0 g (10%) of the monomer pre-emulsion is charged to the reactor followed by 10 g of 5% sodium persulfate solution. The contents exotherm to about 85°C, and after cooling back to 80°C, the addition of the remaining pre-emulsion is started and continued progressively over 2.5 hours until complete. The reactor contents are heated for an additional 30 minutes at 80°C to complete the conversion of monomer to copolymer and then cooled. 35

The product is a low viscosity latex of solids content 29.2%, RVT Brookfield viscosity 26.8 mPa-s (No. 1 spindle at 100 rpm), pH of 2.9, and average particle size of 92 nm. Upon neutralization to pH 9 with ammonium hydroxide of a 1% active concentration of this latex in water, a clear solution is obtained with RVT Brookfield viscosity of 1375 mPas (No. 3 spindle at 10 rpm). 40

45

50

55

60

65

TABLE I
EXAMPLES OF NOVEL URETHANE MONOMERS
REACTANTS

Example No.	Mono- Isocyanate Used	Ethoxylated Surfactant Used	
		Hydrophobe	Ethylene oxide (Moles)
1-A	M-TMI	Nonyl-Phenol	9
1-B	M-TMI	Nonyl-Phenol	15
1-C	M-TMI	Nonyl-Phenol	30
1-D	M-TMI	Nonyl-Phenol	50
1-E	M-TMI	Nonyl-Phenol	100
1-F	M-TMI	Octyl-Phenol	40
1-G	M-TMI	Dinonyl-Phenol	49
1-H	M-TMI	Dinonyl-Phenol	150
1-I	M-TMI	Lauryl (C12)	23
1-J	M-TMI	Stearyl (C18)	30
1-K	M-TMI	Oleyl (C-18)	20
1-L	IEM	Nonyl-Phenol	50

The surfactants used in the above Table are:

<u>Example No.</u>	<u>Trade Name</u>
1-A	Igepal CO-630
1-B	Igepal CO-730
1-C	Igepal CO-880
1-D	Igepal CO-970
1-E	Igepal CO-990
1-F	Igepal CA-890
1-G	Igepal DM-880
1-H	Igepal DM-970
1-I	Siponic L-25
1-J	Siponic E-15
1-K	Emulphor ON-870
1-L	Igepal CO-970

The abbreviations used in the above Table are:

M-TMI = Alpha, alpha-dimethyl-m-isopropenyl benzyl isocyanate

IEM = Isocyanatoethyl methacrylate

Igepal® and Emulphor® are trademarks of GAF Corporation. Siponic® is a trademark of Alcolac.

Table II
PREPARATIONS OF ALKALI-SOLUBLE LATEX THICKENERS
USING EXAMPLE 1-D URETHANE MONOMER

<u>Example</u> (No.)	<u>Thickener Monomer Composition</u>			<u>1% Aqueous</u> <u>Solution</u> <u>Viscosity</u> <u>at PH 9</u>
	<u>Urethane</u> <u>Monomer</u> (Wt. %)	<u>Ethyl</u> <u>Acrylate</u> (Wt. %)	<u>Methacrylic</u> <u>Acid</u> (Wt. %)	(mPa-s)
2-A**	None	25	75	52
2-B**	None	40	60	70
2-C**	None	55	45	96
2-D**	None	70	30	220
3-A	5	45	50	470
3-B	5	50	45	400
3-C	10	35	55	764
3-D	10	40	50	1220
3-E	10	50	40	1375
3-F	10	55	30	1553
3-G	15	25	60	268
3-H	15	35	50	1730
3-I	15	40	45	2025
3-J	15	45	40	2140
3-K	15	55	30	2575
3-L	15	65	20	1416
3-M	15	70	15	178
3-N	20	50	30	4880
3-O	25	30	45	3400
3-P	25	35	40	5680
3-Q	25	40	35	7350
3-R	25	50	25	4400
3-S	30	25	45	5180
3-T	30	40	30	6100
3-U	45	25	30	880
Cellulosic**	--	--	--	2196

TABLE II CONTINUED
Properties In An Interior Flat Paint

<u>Example</u> (No.)	<u>Thickening</u> <u>Efficiency</u> (Dry Lbs.*)	<u>Brush Drag</u> <u>Viscosity</u> (Poise)	<u>Leveling</u> <u>Viscosity</u> (Poise)
2-A**	32.42	1.85	3523
2-B**	20.52	2.17	3555
2-C**	22.48	2.39	2607
2-D**	23.04	2.43	3444
3-A	12.87	2.14	1801
3-B	11.93	2.36	1501
3-C	9.72	2.17	1659
3-D	9.00	2.31	2686
3-E	9.30	1.95	2686
3-F	--	--	--
3-G	11.86	2.17	1975
3-H	8.01	2.45	2038
3-I	8.02	2.27	2449
3-J	7.97	2.10	2054
3-K	8.72	2.15	3729
3-L	13.50	1.30	5609
3-M	22.34	1.29	6636
3-N	7.96	2.10	2054
3-O	--	--	--
3-P	7.16	1.98	1975
3-Q	--	--	--
3-R	7.80	1.46	2528
3-S	6.92	1.53	1248
3-T	7.55	1.29	1375
3-U	8.79	2.43	1122
Cellulosic**	7.00	0.96	2212

*Number of pounds (1 lb = 0.453 kg) of material which must be added to 378.5 l (100 gallons) of latex paint to provide 92-96 KU Stormer Paint Viscosity; 1 Poise = 10^{-1} Pa·s
 Cellulosic is Natrosol® 250 HBR, a hydroxyethyl cellulose from Hercules Inc.

**Comparative

TABLE III
OTHER EXAMPLES OF ALKALI-SOLUBLE THICKENERS
USING URETHANE MONOMERS FROM TABLE I

Thick- ener Example	Urethane Monomer Example Used	Thickener Monomer Composition			1% Aqueous Solution Viscosity at PH 9
		Urethane Monomer	Ethyl Acrylate	Metha- crylic Acid	
(No.)	(No.)	(Wt.%)	(Wt.%)	(Wt. %)	(mPa.s)
4	1-A	10	50	40	417
5	1-A	25	50	25	1272
6	1-A	10	40	50	625
7	1-A	25	35	40	1550
8	1-B	10	50	40	477
9	1-C	10	50	40	969
10	1-E	10	50	40	2256
11	1-F	10	50	40	1480
12	1-G	10	50	40	10053
13	1-G	25	50	25	17160
14	1-G	10	40	50	6680
15	1-G	25	35	40	29600
16	1-H	10	50	40	3080
17	1-I	10	50	40	752
18	1-K	10	50	40	1830
19	1-L	10	50	40	1066

TABLE III CONTINUED

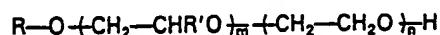
Urthane Properties In An Interior Flat Paint

	Thickener Example	Monomer Example Used	Thickening Efficiency (Dry Lbs*)	Brush Drag Viscosity (Poise)	Leveling Viscosity (Poise)
5					
10	4	1-A	9.56	1.66	3081
	5	1-A	--	--	--
	6	1-A	9.85	1.88	1896
15	7	1-A	7.95	1.05	1722
	8	1-B	9.92	1.75	3160
	9	1-C	8.81	2.25	4266
20	10	1-E	10.25	1.82	6794
	11	1-F	8.09	1.89	1485
	12	1-G	8.52	2.20	2171
	13	1-G	--	--	--
25	14	1-G	6.82	1.31	2007
	15	1-G	4.74	1.15	1896
	16	1-H	7.73	1.47	3444
30	17	1-I	8.52	2.20	2171
	18	1-K	8.80	1.09	3350
	19	1-L	9.45	2.04	1438

* see note in Table II.

Claims

1. A nonionic urethane monomer which is the urethane reaction product of a monohydric nonionic surfactant with a monoethylenically unsaturated monoisocyanate.
2. A nonionic urethane monomer as recited in claim 1 in which said monohydric nonionic surfactant has the formula:



- in which R is an alkyl group containing 6—22 carbon atoms or an alkaryl group containing 8—22 carbon atoms, R' is C₁—C₄ alkyl, n is an average number from about 6—150, and m is an average number of from 0—50 provided n is at least as great as m and n + m = 6—150.

3. A nonionic urethane monomer as recited in claim 2 in which said monohydric nonionic surfactant is an ethoxylate of a nonyl phenol containing about 50 moles of adducted ethylene oxide per mole of nonyl phenol and is reacted with alpha, alpha-dimethyl-m-isopropenyl benzyl isocyanate.

4. An alkali soluble thickener which is an aqueous emulsion copolymer of:
- (A) about 20—70 weight percent of an alpha, beta-monoethylenically unsaturated carboxylic acid,
- (B) about 20—80 weight percent of a monoethylenically unsaturated monomer lacking surfactant capacity;

- (C) about 0.5—60 weight percent of a non-ionic urethane monomer recited in claim 1; and
- (D) from 0 up to about 2 weight percent of a polyethylenically unsaturated monomer.

5. An alkali soluble thickener as recited in claim 4 in which said component (C) is the nonionic urethane monomer recited in claim 3.

6. An alkali soluble thickener as recited in claim 4 in which said component (A) is methacrylic acid present in an amount of from 25—55 weight percent and said component (B) is ethyl acrylate present in an amount of 30—65 weight percent.

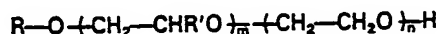
7. An aqueous latex containing the neutralized aqueous emulsion copolymer of claim 4.

8. An aqueous latex as recited in claim 7 in which said copolymer is neutralized with ammonium hydroxide.

9. An aqueous latex containing the neutralized aqueous emulsion copolymer of claim 5.
10. An aqueous latex containing the neutralized aqueous emulsion copolymer of claim 6.

5 Patentansprüche

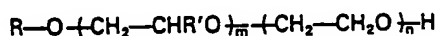
1. Nichtionisches Urethan-Monomer, das das Urethan-Reaktionsprodukt eines einwertigen (monohydrischen) nichtionischen Tensids mit einem monoethylenisch ungesättigten Monoisocyanat darstellt.
10 2. Nichtionisches Monomer nach Anspruch 1, worin das monohydrische nichtionische Tensid die Formel



- 15 hat, worin R eine Alkylgruppe mit 6 bis 22 Kohlenstoffatomen oder eine Alkarylgruppe mit 8 bis 22 Kohlenstoffatomen, R' C₁-C₄-Alkyl, n eine Durchschnittszahl von etwa 6 bis 150 und m eine Durchschnittszahl von 0 bis 50 darstellt, vorausgesetzt, daß n mindestens so groß wie m ist und daß n + m = 6 bis 150.
3. Nichtionisches Monomer nach Anspruch 2, worin das einwertige nichtionische Tensid ein Ethoxylat eines Nonylphenols darstellt, das etwa 50 Mol adduziertes Ethylenoxid je Mol Nonylphenol enthält, und das mit alpha, alpha-Dimethyl-m-isopropenylbenzyl-isocyanat umgesetzt ist.
20 4. Alkalilösliches Verdickungsmittel, das ein wäßriges Emulsions-Copolymer aus:
(A) etwa 20 bis 70 Gew.-% einer alpha, beta-monoethylenisch ungesättigten Carbonsäure,
(B) etwa 20 bis 80 Gew.-% eines monoethylenisch ungesättigten Monomers ohne
25 Tenseideigenschaften,
(C) etwa 0,5 bis 60 Gew.-% eines nichtionischen Urethanmonomers gemäß Anspruch 1; und
(D) 0 bis etwa 2 Gew.-% eines polyethylenisch ungesättigten Monomers darstellt.
5. Alkalilösliches Verdickungsmittel nach Anspruch 4, worin die Komponente (C) das nichtionische Urethanmonomer gemäß Anspruch 3 darstellt.
30 6. Alkalilösliches Verdickungsmittel nach Anspruch 4, worin die Komponente (A) Methacrylsäure darstellt und in einer Menge von 25 bis 55 Gew.-% vorhanden ist und worin die Komponente (B) Äthylacrylat darstellt und in einer Menge von 30 bis 65 Gew.-% vorliegt.
7. Wäßriger Latex, enthaltend das neutralisierte wäßrige Emulsions-Copolymer von Anspruch 4.
8. Wäßriger Latex nach Anspruch 7, worin das Copolymer mit Ammoniumhydroxid neutralisiert ist.
35 9. Wäßriger Latex, enthaltend das neutralisierte wäßrige Emulsions-Copolymer von Anspruch 5.
10. Wäßriger Latex, enthaltend das neutralisierte wäßrige Emulsions-Copolymer von Anspruch 6.

Revendications

- 40 1. Monomère uréthane non ionique qui est le produit uréthane d'une réaction entre un tensioactif non ionique monohydrique et un monoisocyanate à insaturation monoéthylénique.
2. Monomère non ionique selon la revendication 1, dans lequel ledit tensioactif non ionique monohydrique a pour formule:



- dans laquelle R est un groupe alkyle contenant 6—22 atomes de carbone ou un groupe alkylaryle contenant 8—22 atomes de carbone, R' est un groupe alkyle en C₁—C₄, n est un nombre moyen d'environ 6—150 et m est un nombre moyen de 0—50, à condition que n soit au moins aussi grand que m et que n + m = 6—150.
50 3. Monomère non ionique selon la revendication 2, dans lequel ledit tensioactif non ionique monohydrique est un éthoxylat d'un nonylphénol contenant environ 50 moles d'oxyde d'éthylène additionné par mole de nonylphénol et est mis à réagir avec l'alpha, alpha-diméthyl-m-isopropénylbenzylisocyanate.
55 4. Epaississant soluble dans les alcalis qui est un copolymère en émulsion aqueuse de:
(A) environ 20—70% en poids d'un acide carboxylique à insaturation alpha, beta-monoéthylénique,
(B) environ 20—80% en poids d'un monomère à insaturation monoéthylénique n'ayant pas de pouvoir tensioactif,
(C) environ 0,5—60% en poids d'un monomère uréthane non ionique selon la revendication 1, et
60 (D) de 0 jusqu'à environ 2% en poids d'un monomère à insaturation polyéthylénique.
5. Epaississant soluble dans les alcalis selon la revendication 4, dans lequel ledit constituant (C) est le monomère uréthane non ionique selon la revendication 3.
6. Epaississant soluble dans les alcalis selon la revendication 4, dans lequel ledit constituant (A) est un acide méthacrylique présent en une quantité de 25—55% en poids et ledit constituant (B) est l'acrylate d'éthyle, présent en une quantité de 30—65% en poids.

EP 0 173 109 B1

7. Latex aqueux contenant le copolymère en émulsion aqueuse neutralisé selon la revendication 4.
8. Latex aqueux selon la revendication 7, dans lequel ledit copolymère est neutralisé avec de l'hydroxyde d'ammonium.
9. Latex aqueux contenant le copolymère en émulsion aqueuse neutralisé selon la revendication 5.
10. Latex aqueux contenant le copolymère en émulsion aqueuse neutralisé selon la revendication 6.

10

15

20

25

30

35

40

45

50

55

60

65